

#### **Cambridge International Examinations**

Cambridge International General Certificate of Secondary Education

COMBINED SCIENCE 0653/62

Paper 6 Alternative to Practical

May/June 2017

MARK SCHEME
Maximum Mark: 60

#### **Published**

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Question	Answer	Marks
1(a)(i)	quality drawing using at least half the space and not feathery; male parts – anther and filament; female parts – stigma (and ovary); some petals;	4
1(a)(ii)	correctly labelled: anther; stigma;	2
1(b)(i)	2 lines drawn edge to edge ; correct measurement of photograph 47 mm $\pm$ 1 mm <b>AND</b> (sensible) flower measurement (larger than photograph) ;	2
1(b)(ii)	correct calculation ;	1
1(c)	stigma circled (on Fig.1.1);	1

© UCLES 2017 Page 2 of 7

Question	Answer	Marks
2(a)(i)	all values of V correct 7, (10), 14, 21;	1
2(a)(ii)	axes labelled with units; linear scales using at least half of grid in each direction; at least three points plotted correctly within half a small square; best straight line <b>or</b> best curve;	4
2(a)(iii)	the higher the temperature the higher the rate of the reaction ;	1
2(b)(i)	removes timing error associated with starting the stopclock and connecting apparatus / could be too fast in first minute due to powder on chips / air in measuring cylinder;	1
2(b)(ii)	the surface area (of the chips) is reduced / reaction slowed by smaller chips;	1
2(b)(iii)	bubble into water ; count bubbles in a certain time / time for certain number of bubbles ;	max 2
	OR	
	connect delivery tube to a gas syringe ; measure volume in a certain time / time for a certain volume ;	
	OR	
	place reaction flask on a balance ; measure mass in a certain time / time for certain drop in mass ;	

© UCLES 2017 Page 3 of 7

Question	Answer	Marks
3(a)(i)	51.3 (g) ;	1
3(a)(ii)	67 (cm <sup>3</sup> ) ;	1
3(a)(iii)	read to bottom of meniscus / take reading at eye level / perpendicular to scale ;	1
3(a)(iv)	1.03; g/cm <sup>3</sup> ;	2
3(b)(i)	18 (cm <sup>3</sup> ) ;	1
3(b)(ii)	$(\frac{18.1}{(b)(i)})$ 1.0 / 1.01 (g / cm <sup>3</sup> );	2
	2 or 3 significant figures ;	
3(c)(i)	zero error on balance / test-tube touching side of cylinder ;	1
3(c)(ii)	measuring cylinder otherwise wet / contains some water when its 'dry' mass is measured ;	1

© UCLES 2017 Page 4 of 7

Question	Answer	Marks
4(a)	leave in the dark ; at least 24 hours ;	2
4(b)(i)	alcohol and warm (to take out chlorophyll); iodine (solution); (brown) to blue-black;	3
4(b)(ii)	Benedicts (solution); heat; red (most sugar) / orange / yellow (less) / green (little);	3
4(b)(iii)	no naked flame ethanol flammable / use water-bath with ethanol as ethanol flammable / goggles chemicals in eyes / don't touch hot apparatus will burn hands / heatproof gloves so don't burn hands / gloves stop chemical burns ;	1
4(c)	light AND carbon dioxide needed (for photosynthesis);	1

© UCLES 2017 Page 5 of 7

Question	Answer	Marks
5(a)(i)	solid in beaker / solution / mixture ;	1
5(a)(ii)	make sure that all the acid has reacted;	1
5(a)(iii)	filtration;	1
5(b)(i)	(all water gone) decomposes (to copper oxide);	1
5(b)(ii)	evaporate some of the water / leave to evaporate / heat / evaporate ; leave to crystallise / cool ; filter ; leave to dry / press between filter paper ;	max 3
5(c)	add barium nitrate / barium chloride AND white ppt. ;	1
5(d)	zinc oxide ; hydrochloric acid ;	2

© UCLES 2017 Page 6 of 7

Question	Answer	Marks
6(a)(i)	pipette / syringe ;	1
6(a)(ii)	same surface (area) / so same (rate of) evaporation ;	1
6(b)(i)	27.5 (°C); 14.0 (°C);	2
6(b)(ii)	17.8 (°C); 9. <u>0</u> (°C);	2
6(c)(i)	180 (s);	1
6(c)(ii)	cotton wool dropped off / misread thermometer ;	1
6(d)	cotton wool almost dry / rate of evaporation slower / most alcohol evaporated / all alcohol evaporated ;	1
6(e)	reasonable sketch (left to right curve, starting high on LHS);	1

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